



# **Behavior of Micro-Stickies in Papermaking Environment**

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# Outline



- **Background of micro-stickies**
- **Objectives of research**
- **Experiments & Discussion**
  - Charge determination of model micro-stickies
  - Stability of micro-stickies (no fibers)
  - Interactions between micro-stickies and fibers
- **Conclusions**
- **Future work**

# Micro-Stickies

- What are micro-stickies?
  - Can pass through 0.006" (about 125 $\mu$ m) slotted screen
  - Defined as a size less than 100  $\mu$ m
- Problems associated with micro-stickies
  - Can agglomerate and deposit on the wires and felts
  - Affect the paper properties and appearance



# What Affects the Formation of Micro-Stickies ?



- Mechanical action in the pulper or other equipment.
- Materials that detackify adhesive contaminants promote the formation of micro-stickies.
- Some adhesives applied to the paper are already in a small, discrete size

# Micro-Stickies Previous Research

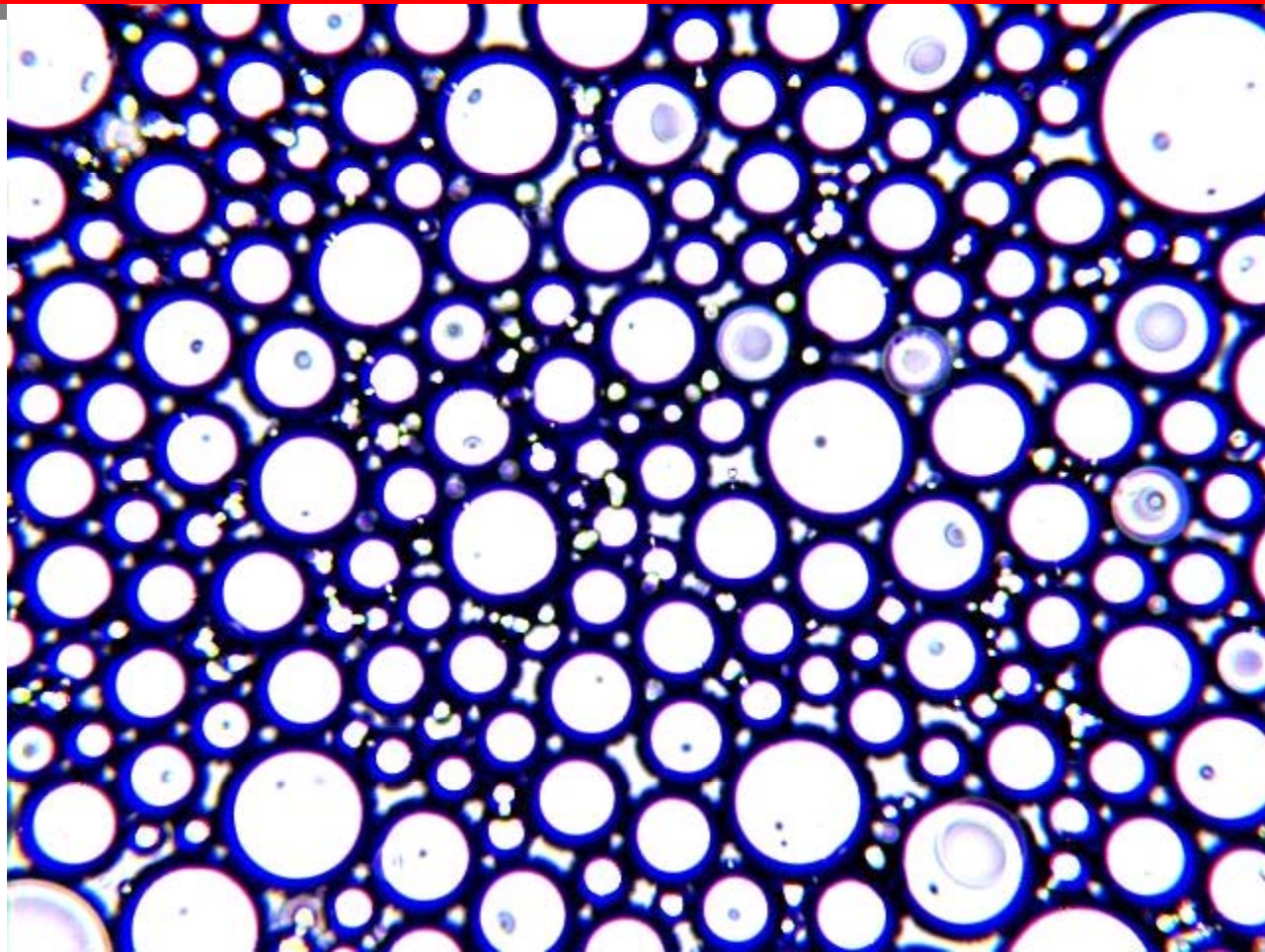
- **R.Klein and H. Grossmann (1997)**  
Discussed difficulty in the measurements and investigated the effects of cationic polymers on different micro-stickies
- **I. M. Hutten and Sujit Banerjee (1997)**  
Micro-stickies will agglomerate in pure water but fibers inhibit this process
- **B. Carre and J. Brun (1996)**  
Maximum deposition of micro-stickies occurred at charge neutral point of the entire system
- **A. Setter (1993)**  
Micro-stickies can be removed from white water by precipitation or flocculation by cationic polymer followed by micro-flotation

# Objectives:



- Develop a model system and measurement methods to study the destabilization of micro-stickies
- Determine the effect that salts, poly-electrolytes and fibers have on the behavior of micro-stickies

# Acrylic Microsphere Adhesive Emulsion: Size Characteristic



Magnification 50X10X0.6

15  $\mu\text{m}$

# Charge Demand Characteristic:

- Micro-sticky            -18  $\mu\text{eq}/\text{gram}$
- Fiber                      -10  $\mu\text{eq}/\text{gram}$
- Cationic Starch        +220  $\mu\text{eq}/\text{gram}$

# Stability of Model Micro-Stickies



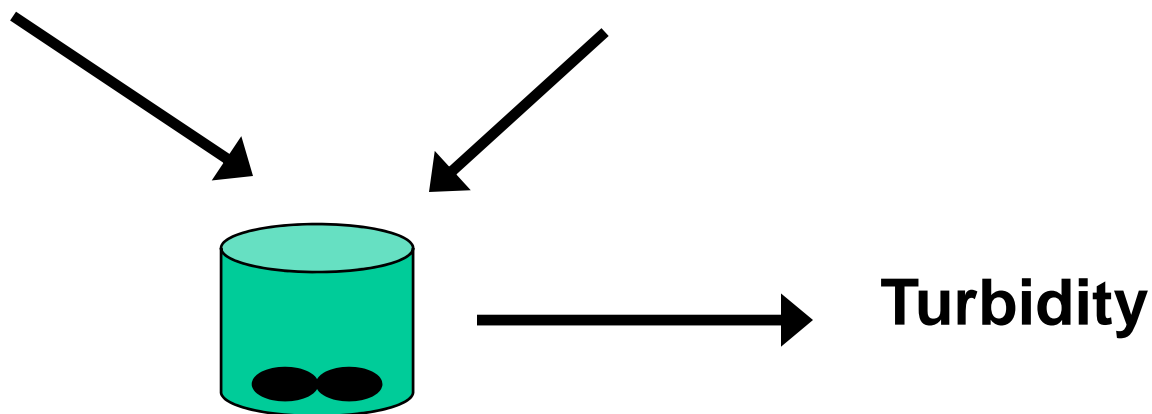
Question:

- How will different cationic polymers will affect the stability of the micro-sticky?

# Experimental Procedure

**Cationic Polymer**

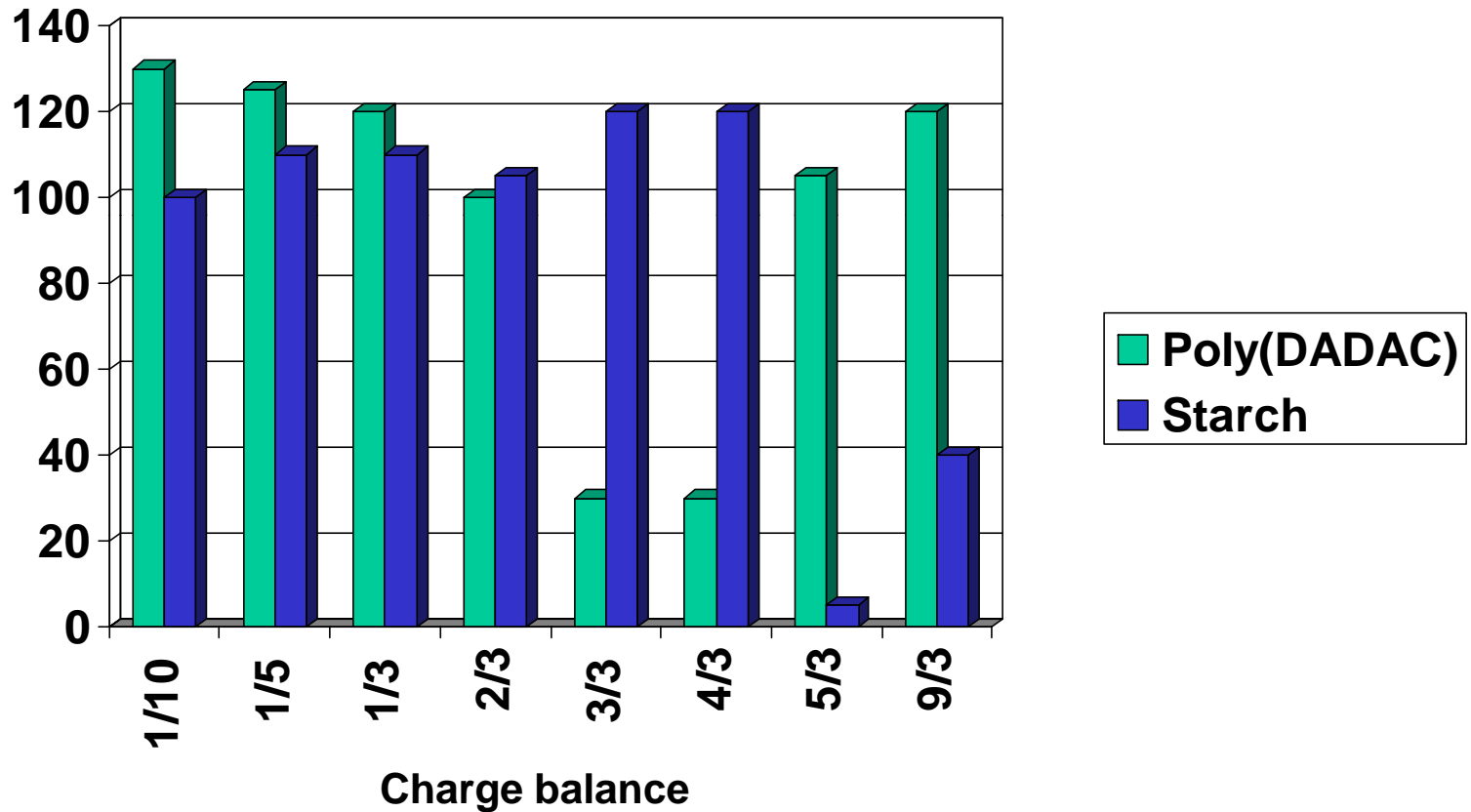
**MICRO STICKIES**



**Mix for 15 min.**

# Effect of Polymers on Stability of Micro-stickies

## Turbidity Overnight (NTU)



# Effect of Polymers on Stability of Micro-stickies



- Poly(DADMAC) was more effective in destabilizing the micro-sticky
- Poly(DADMAC) created strong flocs whereas the starch created weak flocs
- Maximum flocculation for
  - Poly(DADMAC) at neutral point
  - Cationic starch much greater than neutral point
- Steric and charge stabilization for excess amounts of cat material

# Stability of Model Micro-Stickies



Question:

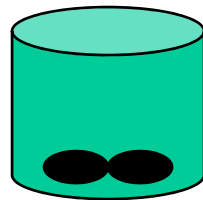
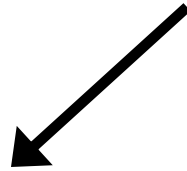
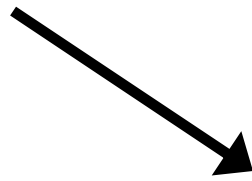
- How will salt concentration and valence of counter-ion affect the stability of the micro-sticky?

# Experimental Procedure

**Salt:**

**NaCl, CaCl<sub>2</sub>, or Alum**

**MICRO STICKIES**



**Turbidity**

**Mix for 15 min.**

# Critical Coagulation Concentration (CCC)

Schultz-Hardy rule

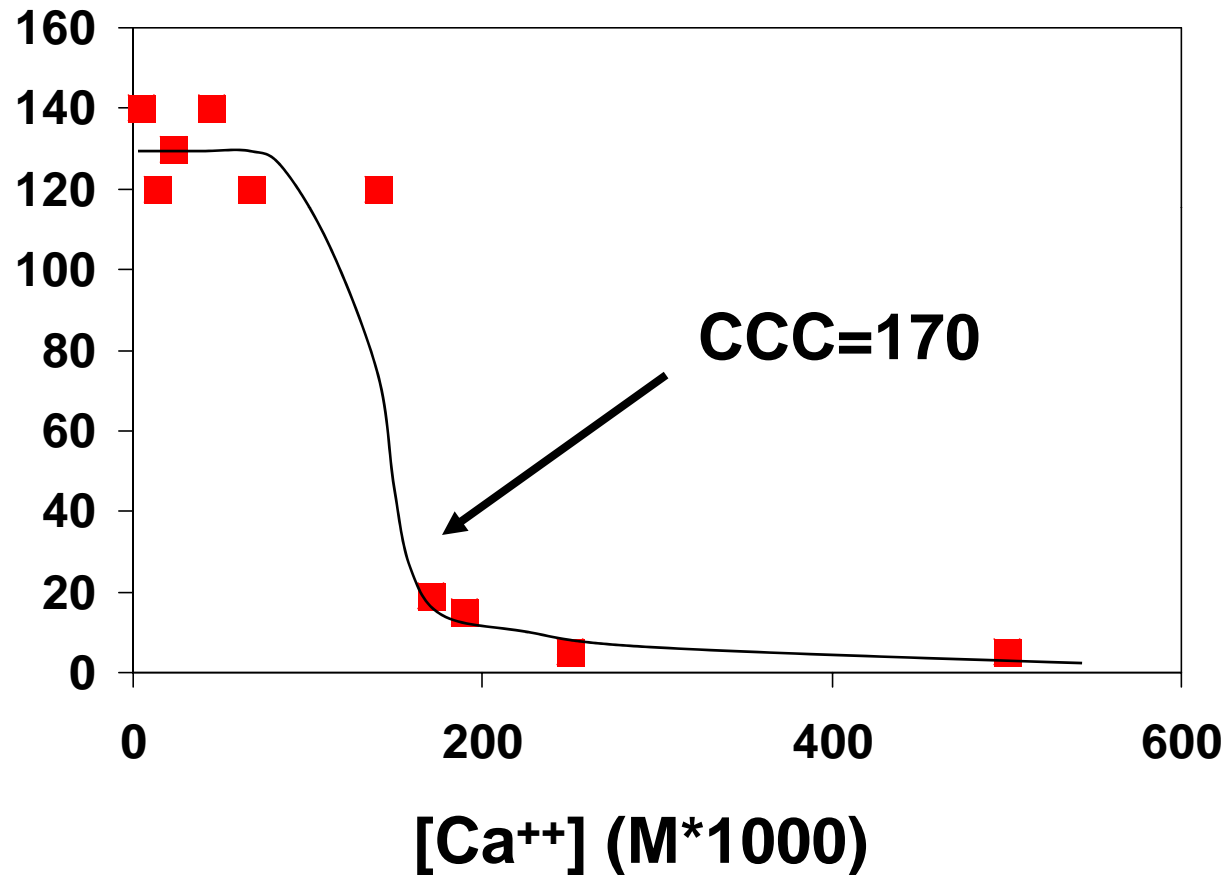
$$\text{CCC} \propto z^{-6}$$

Valency (Z)	CCC
1	1
2	$0.016 = 1/2^6$
3	$0.0014 = 1/3^6$

Double-layer thickness depends on counter-ion valency

# Effect of $\text{Ca}^{++}$ on Micro-Sticky

Turbidity (NTU)



# Effect of Salts on Stability of Micro-Sticky

	CCC	
	Observed	Expected
Na <sup>+</sup>	1.3 M	--
Ca <sup>++</sup>	1.7E-2	2.0E-2
Al <sup>+++</sup>	1.3E-3	1.8E-3

# Effect of Salts on Stability of Micro-Sticky

- The coagulation of the micro-sticky closely follows the behavior of the Shultz-Hardy Rule
- Comparison of mill data in the literature to these results indicates:
  - **[Ca<sup>++</sup>] commonly exists below the CCC**
  - **[Al<sup>+++</sup>] commonly exists around the CCC**
- **MICROSTICKIES BEHAVE AS COLLOIDS WITH AN ELECTROSTATIC DOUBLE LAYER PREVENTING CONTACT**

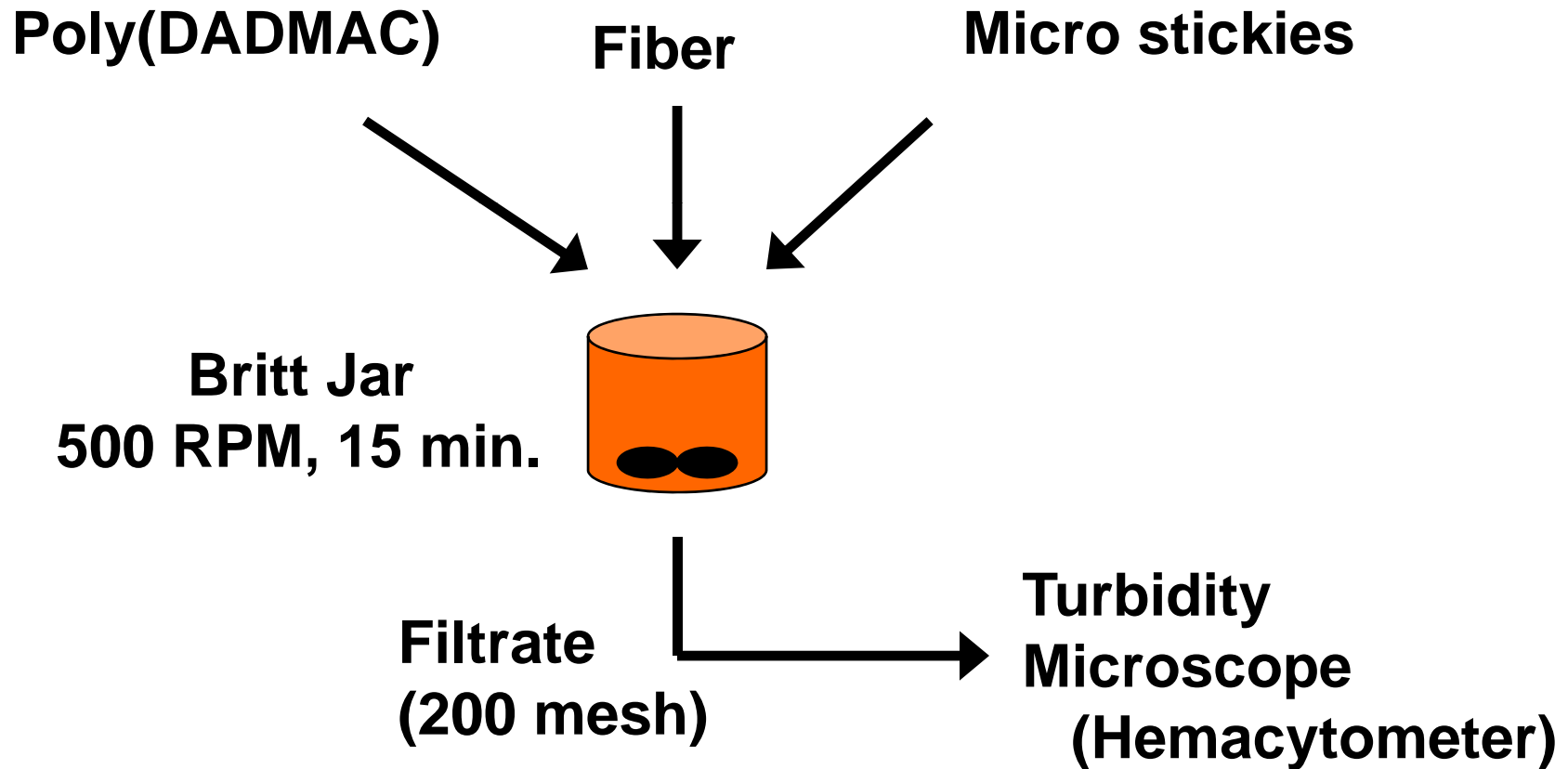
# Stability of Model Micro-Stickies



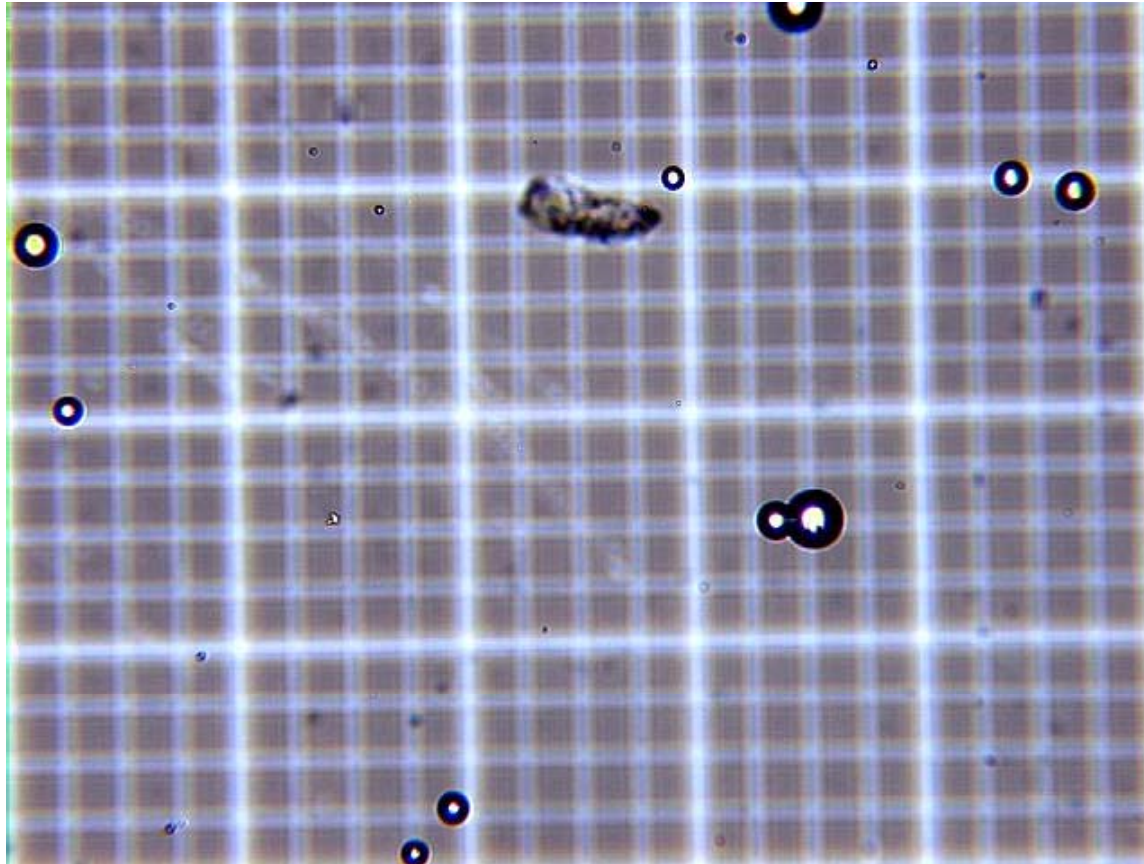
Question:

- How will fibers affect the stability of the micro-sticky in the presence of a cationic polyelectrolyte?

# Experimental Procedure

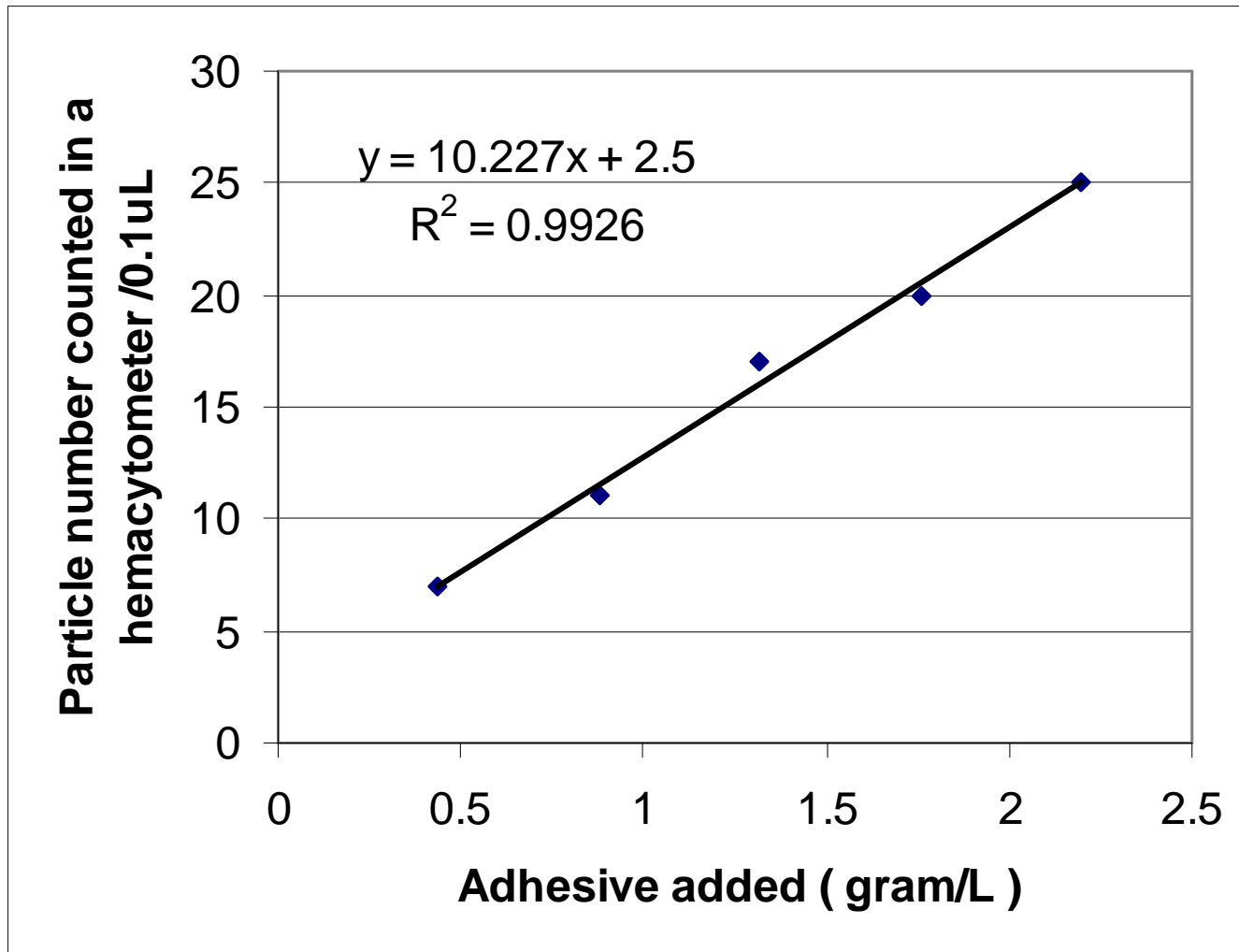


# Micro-Sticky Particle Identification with Hemacytometer

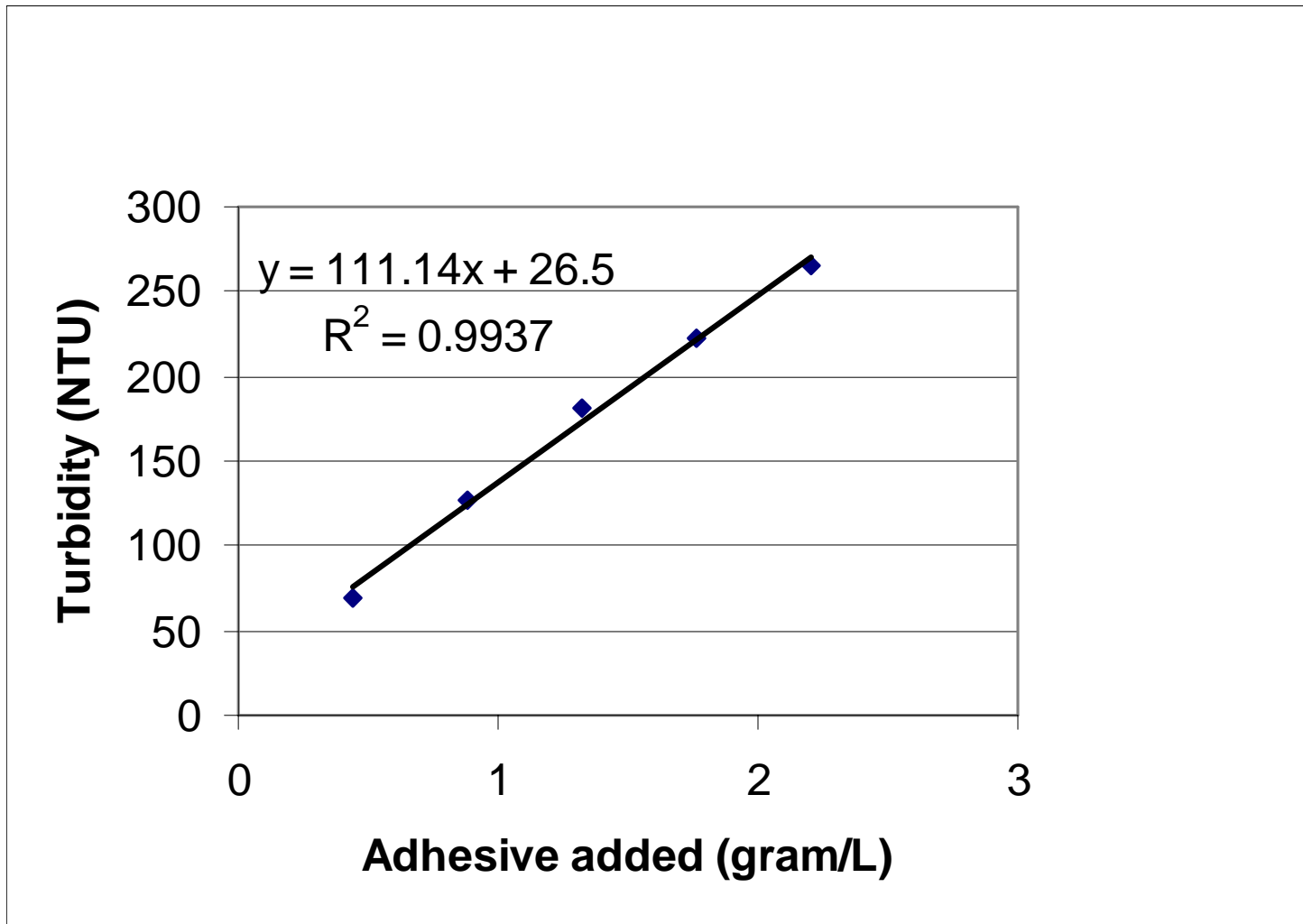


0.2 mm

# Calibration Line for Hemacytometer



# Calibration Line for Turbidity

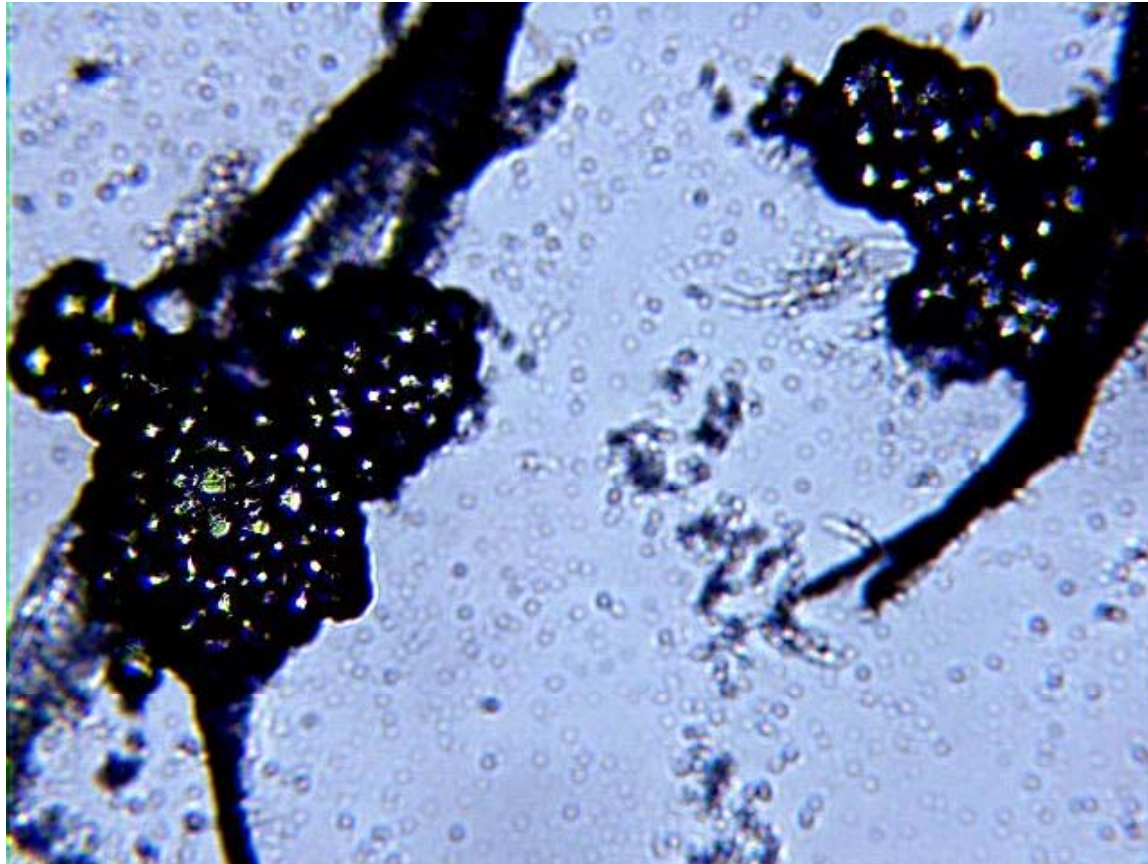


# Fibers and Micro-Stickies (No Poly(DADMAC))



Magnification 10X10X0.6

# Fibers and Micro-Stickies (with Poly(DADMAC))



Magnification 10X10X0.6

# Effect of Fibers on Micro-Stickies in Filtrate

<b>Fiber</b>	<b>Sticky</b>	<b>DADMAC</b>	<b>Turbidity</b>	<b>Count</b>
-	+	-	170	18
+	+	-	136	8
+	+	+(1)	54	0
+	+	+(2)	7	0

**(1) to neutralize adhesive only**

**(2) to neutralize adhesive + fiber**

# Effect of Fibers on Micro-Stickies



- Fibers compete with the micro-stickies for the poly(DADMAC)
- Micro-stickies flocculate and attach to fibers in the presence of poly(DADMAC)
- Turbidity measurements and the Hemacytometer were useful in tracking the fate of the micro-stickies

# Conclusions



- Poly(DADMAC) and cationic starch show optimum points in destabilizing the micro-sticky
- The micro-sticky closely follows the Shultz-Hardy rule for critical coagulation concentration
- The presence of fibers can compete for cationic species with the micro-sticky